# FOUR YEAR UNDERGRADUATE PROGRAM(2024 – 28) DEPARTMENT OF INDUSTRIAL CHEMISTRY COURSE CURRICULUM

P	ART-A: Introdu	ction			
Program: Bachelor in Science (Degree/Honors)		Semester VI	Session: 2024-2025		
1	Course Code	ICSE-04T			2
2	Course Title	1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2			-
3	Course Type		DSE		
4	Pre-requisite(if,any)				
5	Course Learning. Outcomes(CLO)  To address the student about the inorganic materials which is imposing in industries.  To understand the preparation, type and use of silicates.  To understand the types of fertilizer as inorganic compound.  To understand the alloy formation and batteries in industries.			portai	
6	Credit Value	3 Credits	Credit = 15 Hou	rs -learning & Observation	on -
7	Total Marks	Max. Marks:	100	Min Passing Marks:40	
A	RT -B: Content	of the Cou	rse		
	Total No. of Tea	ching-learning 1	Periods(01 Hr. per periods)	od) - 45 Periods (45 Hour	·s)
Unit		Topics(Course contents)			No.o Perio
1	Glass: Glassy state Manufacture and proof glasses: Soda lin fluorosilicate, colou Ceramics: Importar technology ceramics fullerenes, carbon na	ocessing of glass.  ne glass, lead glasted glass, photose  nt clays and felds  s and their applica	Composition and proper ss, armoured glass, safet nsitive glass. spar, ceramic, their type ations, superconducting a	e and non-silicate glasses).  rties of the following types ty glass, borosilicate glass,  es and manufacture. High and semiconducting oxides,	12
II Fertilizers:  Different types of fertilizers. Manufacture of the following fertilizers: Urea, ammonium nitrate, calcium ammonium nitrate, ammonium phosphates; polyphosphate superphosphate, compound and mixed fertilizers, potassium chloride, potassium sulphate			11		
II	Alloys Classification of alloys, ferrous and non-ferrous alloys, Specific properties of elements in alloys. Manufacture of Steel (removal of silicon decarbonization, demanganization, desulphurization dephosphorisation) and surface treatment (argon treatment, heat treatment, nitriding, carburizing). Composition and properties of different types of steels.				11
IV	V Batteries Primary and secondary batteries, battery components and their role, Characteristics of Battery. Working of following batteries: Pb acid, Li-Battery, Solid state electrolyte battery. Fuel cells, Solar cell and polymer cell.				

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Silving (1)

## PART-C:Learning Resources

### Text Books, Reference Books and Others

### Text Books Recommended -

- 1. Jain, P. C., & Jain, M. (2010. Engineering chemistry. Dhanpat Rai & Sons.
- 2. Gopalan, R., Venkappayya, D., & Nagarajan, S. (2016). Engineering chemistry. Vikas Publishing House Pvt Ltd.
- 3. Sharma, B. K. (2005). Engineering chemistry. Goel Publishing House.

### Reference Books Recommended -

- 1. Stocchi, E. (Vol. 1). Industrial chemistry. Ellis Horwood Ltd.
- 2. Felder, R. M., & Rousseau, R. W. (2007). Elementary principles of chemical processes, Wiley.
- 3. Kingery, W. D., Bowen, H. K., & Uhlmann, D. R. (2008). Introduction to ceramics, Wiley India Pvt. Ltd.
- 4. Kent, J. A. (Ed.). (1993). Riegel's handbook of industrial chemistry. CBS Publishers & Distributors.

### Online Resources:

- https://www.usgs.gov/centers/national-minerals-information-center/silica-statistics-and-information -
- https://www.fertilizer.org/ https://www.asminternational.org/ -
- https://www.nrel.gov/ -
- https://www.energy.gov/batteries

<b>PART-D:Assessment</b>	andEvaluation
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**Suggested Continuous Evaluation Methods:** 

Maximum Marks:

100 Marks

Continuous Internal Assessment(CIA):30 Marks

End Semester Exam(ESE):70 Marks

End Schiester Exam(E	SE). 70 IVIAI KS			
Continuous Internal	Internal Test / Quiz-(2): 20 +20	Better marks out of the two Test / Quiz		
Assessment(CIA):	Assignment/Seminar- 10	obtained marks in Assignment shall		
(By Course Teacher)	Total Marks -30	considered against 30 Marks		
<b>End Semester</b>	Two section – A & B			
Exam (ESE):	Section A: Q1. Objective $-10 \text{ x1} = 10 \text{ Mark}$ ; Q2. Short answer type-			
and the second s	=20Marks			
	Section B: Descriptive answer type q	ts., 1out of 2 from each unit-		

Name and Signature of Convener & Members of CBoS:

Jul Ball

4x10=40Marks

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# FOUR YEAR UNDERGRADUATE PROGRAM(2024 – 28) DEPARTMENT OF INDUSTRIAL CHEMISTRY COURSE CURRICULUM

		COURS	E CORRIGEZON		
P	ART-A: Introd	uction			
Program:Bachelor in Science (Degree/Honors)		Semester VI	Session: <b>2024-20</b>	25	
1	CourseCode	seCode ICSE-04P			
2	CourseTitle	INORGANIC N LAB COURSE	INORGANIC MATERIALS OF INDUSTRIAL IMPORTANCE LAB COURSE		
3	CourseType	DSE			
4	Pre-requisite(if,any	)	As per program		
5	Course Learning. Outcomes(CLO)	comes(CLO)  To perform the sample analysis of ore to find the metal percentage.  To demonstrate the metallic coating on ceramics.			
6	CreditValue				
7 TotalMarks		Max.Marks:	50	Min Passing Marks:20	
M	[odule		Topics(Courseconten	ts)	No.off eriod
La Tr Exp	b./Field   1.Determination of   2. Estimation of   2. Estimation of   3. Estimation of   4. Electroless of   5. Determination of   5. Determination   6. Analysis of   7. Preparation   7.	of Calcium in Calcof phosphoric acide metallic coatings of composition	a ammonium sulphate ferti ium ammonium nitrate fer in superphosphate fertilize n ceramic and plastic mate of dolomite (by complexo exide).	unzer. er. erial.	30
K	Teywords Fertilizer, a	lldy, plastic, cemer	nt, zinc oxide .	My Rahua G	1)

### PART-C: Learning Resources

Text Books, Reference Books and Others

### Text Books Recommended -

- 1. Sharma, S. K., & Sharma, R. K. (2018). Practical manual for fertilizer analysis. Directorate of Agricultural Research.
- 2. Sinha, S. K. (1972). A handbook of analysis of soil and plant products. Oxford & IBH Publishing Company.
- 3. Sparks, D. L. (1996). Physico-chemical and biological methods for soil analysis, Kluwer Academic Publishers.

### Reference Books Recommended -

- 1. Stocchi, E. (Vol. 1). Industrial chemistry. Ellis Horwood Ltd.
- 2. Felder, R. M., & Rousseau, R. W. (2007). Elementary principles of chemical processes, Wiley.
- 3. Kingery, W. D., Bowen, H. K., & Uhlmann, D. R. (2008). Introduction to ceramics, Wiley India Pvt. Ltd.
- 4. Kent, J. A. (Ed.). (1993). Riegel's handbook of industrial chemistry, CBS Publishers & Distributors.

#### Online Resources-

- https://echa.europa.eu/substance-information/-/substanceinfo/100.029.076
- https://pubs.acs.org/doi/abs/10.1021/acs.jchemed.2c00764
- https://www.sciencedirect.com/science/article/abs/pii/S0016706121002883
- : https://pubs.acs.org/doi/abs/10.1021/acs.ichemed.2c00764
- o : https://www.astm.org/
- https://www.sciencedirect.com/science/article/abs/pii/S0272884218302165
- https://pubs.acs.org/doi/10.1021/acsreagents.2001

### PART-D:Assessment andEvaluation

**Suggested Continuous Evaluation Methods:** 

Maximum Marks: 50 Marks

Continuous Internal Assessment(CIA):15 Marks

End Semester Exam(ESE):35Marks				
Continuous Internal	Internal Test / Quiz-(2): 10 × 10	Better marks out of the	two Test / Quiz	
Assessment(CIA): (By Course Teacher)	Assignment/Seminar +Attendance- 05 Total Marks -15	+obtained marks in Ass considered agains	_	
End Semester Exam (ESE):	Laboratory / Field Skill Performan J. Performed the Task based on Marks K. Spotting based on tools& tech Marks	lab. work - 20	Managed by Course teacher as per lab. status	
	L. Viva-voce (based on principle/technology) - 05 Marks			

Name and Signature of Convener & Members of CBoS:

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